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Key indicators

Single-crystal X-ray study T = 173 K Mean σ (C–C) = 0.002 Å R factor = 0.049 wR factor = 0.129 Data-to-parameter ratio = 15.4

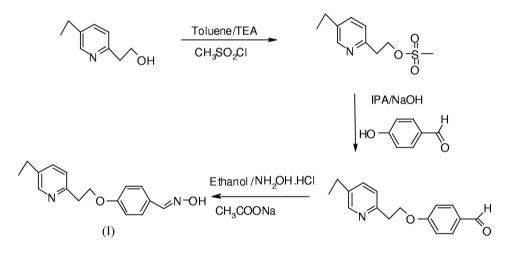
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

4-[2-(5-Ethylpyridin-2-yl)ethoxy]benzaldehyde oxime

The title compound, $C_{16}H_{18}N_2O_2$, is a key intermediate for the synthesis of biologically active novel heterocycles, *e.g.* isoxazolines and isoxazoles. Geometric parameters of the molecule are in the usual ranges. The crystal packing is stabilized by an $O-H \cdots N$ hydrogen bond.

Comment

The title compound, (I), is a key intermediate for the synthesis of biologically active novel heterocycles, *e.g.* isoxazolines, isoxazoles, *etc.* (Hassner & Rai, 1989; Ajay Kumar *et al.*, 2001). A crystal structure determination was carried out in order to elucidate the molecular conformation.



A perspective view of (I) is shown in Fig. 1. Bond lengths and angles can be regarded as normal (Cambridge Structural Database Version 5.26; *ConQuest* Version 1.7; *MOGUL* Version 1.0.1; Allen, 2002). The oxime group is almost coplanar with the benzene ring to which it is attached $[O1-N1-C1-C11 = 179.02 (14)^\circ]$. The ethyl group is far from perpendicular to the pyridyl ring [C5-C4-C24-C23 =74.1 (2)°]. The angle between the benzene and pyridyl rings is 74.80 (5)°. The O2-C2 and C2-C3 bonds are in an antiperiplanar conformation. The O2-C2 bond is nearly coplanar with the benzene ring, whereas the C2-C3 bond deviates markedly from the plane of the pyridyl ring (Table 1). The crystal packing is stabilized by an O-H···N hydrogen bond (Table 2).

Experimental

© 2005 International Union of Crystallography Printed in Great Britain – all rights reserved 2-(5-Ethylpyridin-2-yl)ethanol (1.51 g, 10 mmol) in toluene (10 ml) and triethylamine (2.0 g, 20 mmol) were treated with methane-

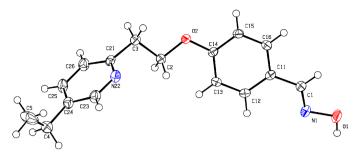


Figure 1

Perspective view of the title compound with the atom numbering; displacement ellipsoids are drawn at the 50% probability level.

sulfonyl chloride (1.14 g, 10 mmol) at 273-278 K. The resulting product was refluxed with NaOH flakes (0.50 g, 12 mmol) and p-hydroxybenzaldehyde (1.22 g, 10 mmol) in isopropyl alcohol (10 ml) to yield 4-[2-(5-ethylpyridin-2-yl)ethoxy]benzaldehyde, which was warmed with NH₂OH·HCl (0.7 g, 10 mmol) and sodium acetate (0.82 g, 10 mmol) in ethanol (10 ml) to produce the title compound, which was crystallized from ethyl acetate (m.p. 391 K) (see scheme). IR (KBr, v cm-1): 3212 (br), 3022 (m), 2930 (m), 2910 (w), 1980 (w), 1658 (m), 1628 (w), 1390 (m), 990 (m), 721 (m). ¹H NMR (CDCl₃, p.p.m.): 1.32 (t, 3H, CH₃), 2.64 (q, 2H, CH₂), 3.21 (t, 2H, CH₂), 4.42 (t, 2H, CH₂), 6.92 (d, 2H, ArH), 7.19 (d, 1H, ArH), 7.42 (d, 2H, ArH), 7.59 (d, 1H, ArH), 8.12 (s, 1H, ArH), 8.54 (s, 1H, CH), 9.88 (s, 1H, OH). ¹³C NMR (CDCl₃, p.p.m.): 15.8 (q, CH₃), 28.1 (t, CH₂), 35.7 (t, CH₂), 73.2 (t, CH₂), 112.2 (d, ArC), 119.2 (d, ArC), 121.0 (s, ArC), 128.4 (d, ArC), 134.3 (d, ArC), 135.6 (s, ArC), 147.8 (d, ArC), 152.2 (d, CH), 161.0 (s, ArC), 162.2 (s, ArC). Analysis calculated for C16H18N2O2: C 71.09, H 6.71, N 10.36%; found: C 71.15, H 6.75, N 10.31%.

Crystal data

$D_x = 1.254 \text{ Mg m}^{-3}$
Mo $K\alpha$ radiation
Cell parameters from 6268
reflections
$\theta = 3.7-26.3^{\circ}$
$\mu = 0.08 \text{ mm}^{-1}$
T = 173 (2) K
Needle, colourless
$0.42 \times 0.13 \times 0.08 \text{ mm}$

Data collection

2222 reflections with $I > 2\sigma(I)$ Stoe IPDS-II two-circle diffractometer $R_{int} = 0.047$ $\theta_{\rm max} = 26.4^{\circ}$ ω scans Absorption correction: none $h = -19 \rightarrow 20$ 7184 measured reflections $k = -6 \rightarrow 6$ 2848 independent reflections $l = -22 \rightarrow 14$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.049$	$w = 1/[\sigma^2(F_o^2) + (0.0771P)^2 + 0.1912P]$
$wR(F^2) = 0.129$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.04 2848 reflections	$(\Delta/\sigma)_{\text{max}} < 0.001$ $\Delta\rho_{\text{max}} = 0.31 \text{ e} \text{ Å}^{-3}$
185 parameters	$\Delta \rho_{\rm min} = -0.30 \text{ e } \text{\AA}^{-3}$
H atoms treated by a mixture of independent and constrained refinement	

Table 1

Selected	geometric	parameters	(Å,	°)	•
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O1-N1	1.4077 (18)	N1-C1	1.284 (2)
C1-N1-O1	110.52 (13)	N1-C1-C11	121.91 (14)
C14-O2-C2-C3 O2-C2-C3-C21	170.10 (13) -169.22 (13)	C2-O2-C14-C15 C2-C3-C21-N22	-164.93 (14) 73.96 (18)

Table 2 Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$\overline{O1\!-\!H1O\!\cdot\cdot\!N22^i}$	0.95 (3)	1.86 (3)	2.7948 (19)	169 (3)
Symmetry code: (i) 1	$-x, y - \frac{1}{2}, \frac{1}{2} - z.$			

All H atoms were located in a difference map. Those bonded to C atoms were positioned geometrically and refined with fixed individual displacement parameters (set to 1.2 times U_{eq} of the parent atom, or 1.5 for methyl groups), using a riding model, with C-H distances ranging from 0.95 to 0.99 Å. The hydroxyl H atom was refined independently with an isotropic displacement parameter.

Data collection: X-AREA (Stoe & Cie, 2001); cell refinement: X-AREA; data reduction: X-AREA; program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: SHELXL97 and PLATON.

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References

- Ajay Kumar, K., Rai, K. M. L. & Umesha, K. B. (2001). J. Chem. Res. (S), pp. 436-438
- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Hassner, A. & Rai, K. M. L. (1989). Synthesis, pp. 57-59.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Stoe & Cie (2001). X-AREA. Stoe & Cie, Darmstadt, Germany.

Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.